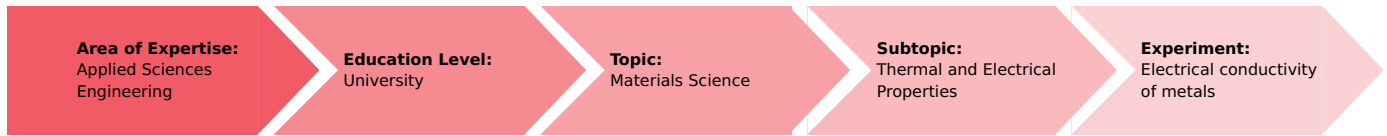


# Electrical conductivity of metals (Item No.: P2350205)

## Curricular Relevance



### Difficulty



Easy

### Preparation Time



1 Hour

### Execution Time



1 Hour

### Recommended Group Size



2 Students

### Additional Requirements:

### Experiment Variations:

- P2350200 "Thermal and electrical conductivity of metals" to show the relation between thermal and electrical conductivity following the Wiedemann-Franz law
- P2350201 "Thermal conductivity of metals"

### Keywords:

Electrical conductivity, Resistivity, Four-point method

## Overview

### Short description

#### Principle

The electrical conductivity of copper and aluminium is determined. For a given thermal conductivity of the materials, the Wiedemann-Franz law is tested.

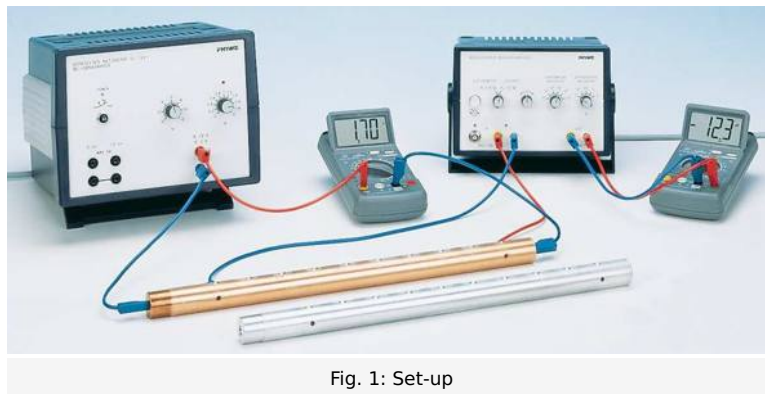


Fig. 1: Set-up

## Equipment

Position No.	Material	Order No.	Quantity
1	PHYWE multitap transformer DC: 2/4/6/8/10/12 V, 5 A / AC: 2/4/6/8/10/12/14 V, 5 A	13533-93	1
2	PHYWE Universal measuring amplifier	13626-93	1
3	Heat conductivity rod, Cu	04518-11	1
4	Heat conductivity rod, Al	04518-12	1
5	Rheostat, 10 Ohm , 5.7A	06110-02	1
6	Digital multimeter 2005	07129-00	2
7	Connecting cord, 32 A, 500 mm, red	07361-01	3
8	Connecting cord, 32 A, 500 mm, blue	07361-04	4
9	Connecting cord, 32 A, 1000 mm, red	07363-01	1

## Tasks

Determine the electrical conductivity of copper and aluminium by recording a current-voltage characteristic line.

Confirm the Wiedemann-Franz law.

## Set-up and procedure

Measurement of the electrical conductivity.

- Set-up the experiment as shown in Fig. 1 and perform the experiment according to the circuit diagram in Fig. 2 (set-up in accordance with a 4-conductor measuring method).

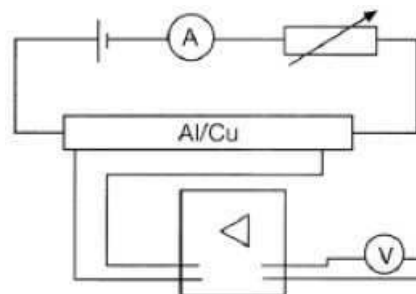


Fig. 2: Circuit diagram.

- Set the voltage on the variable transformer to 6 V.
- The amplifier must be calibrated to 0 in a voltage-free state to avoid a collapse of the output voltage.
- Select the amplifier settings as follows:  
Input: Low Drift  
Amplification:  $10^4$   
Time Constant: 0
- Set the rheostat to its maximum value and slowly decrease the value during the experiment.
- Read and note the values for current and voltage.
- The resistance, and thus the electrical conductivity, can be determined from the measured values.

## Theory and evaluation

The electrical conductivity  $\sigma$  is determined by the resistance  $R$  of the rod and its geometric dimensions ( $l = 0.315 \text{ m}$ ,  $A = 4.91 \cdot 10^{-4} \text{ m}^2$ ):

$$\sigma = \frac{l}{A \cdot R} \quad (1)$$

At room temperature, the conduction electrons in metal have a much greater mean free path than the phonons. For this reason heat conduction in metal is primarily due to the electrons. The resulting correlation between the thermal conductivity  $\lambda$  and the electrical conductivity  $\sigma$  is established by the Wiedemann-Franz law:

$$\frac{\lambda}{\sigma} = LT \quad (2)$$

The Lorenz number  $L$ , which can be experimentally determined using Equation (2), is established by the theory of electron vapour (for temperatures above the Debye temperature) to be:

$$L = \frac{\pi^2}{3} \cdot \left( \frac{k^2}{e^2} \right) = 2.4 \cdot 10^{-8} \frac{\text{W}\Omega}{\text{K}^2} \quad (3)$$

with

$k = \text{Universal gas constant} = 1.38 \cdot 10^{-23} \text{ J/K}$

$e = \text{Elementary unit charge} = 1.602 \cdot 10^{-19} \text{ AS}$

From equation (2), the following values result for  $T = 300 \text{ K}$  and the given thermal conductivities  $\lambda$ :

	$R/10^{-6} \Omega$	$\sigma/10^7 (\Omega\text{m})^{-1}$	$L/10^{-8} \text{ W}\Omega \text{ K}^{-2}$
Al	19.6	3.27	2.24
Cu	12.04	5.33	2.40

with the literature values

$\lambda_{\text{Al}} = 220 \text{ W/(K m)}$

$\lambda_{\text{Cu}} = 384 \text{ W/(K m)}$

The Debye temperatures of copper and aluminium are 335 K and 419 K, respectively. Below the Debye temperature the ratio of the conductivity is smaller than given by Equation (2).